

## Chemical Formulas and Names

A	C	E	T	A	H	P	S	O	H	P	M	U	I	S	S	A	T	O	P	D
S	E	T	A	H	P	S	O	H	P	R	E	V	L	I	S	M	S	F	G	A
I	E	D	I	X	O	R	D	Y	H	N	O	R	I	J	L	A	O	N	P	R
L	A	L	E	A	D	P	H	O	S	P	H	A	T	E	T	G	D	V	X	E
V	M	Z	A	C	E	G	I	K	M	O	I	Q	H	Z	S	N	I	U	W	T
E	M	Y	C	O	P	P	E	R	B	R	O	M	I	D	E	E	U	E	B	A
R	O	D	F	H	T	L	N	P	O	R	T	N	U	V	X	S	M	D	Z	H
H	N	A	C	F	G	I	K	N	M	O	C	Q	M	W	U	I	H	I	W	P
Y	I	Y	B	D	F	H	C	J	L	P	N	P	O	R	T	U	Y	R	V	S
D	U	X	Z	A	E	H	C	G	H	I	K	M	X	O	Q	M	D	O	S	O
R	M	U	W	Y	L	B	D	O	F	H	J	L	I	N	P	H	R	L	R	H
O	O	T	U	O	W	Y	S	A	E	C	G	I	D	K	M	Y	O	H	O	P
X	X	Q	R	S	U	P	W	Y	B	D	F	H	E	T	L	D	X	C	N	M
I	I	I	P	R	H	T	V	X	Z	A	E	C	G	I	K	R	I	N	M	U
D	D	P	R	A	L	U	M	I	N	U	M	B	R	O	M	I	D	E	T	I
E	E	V	T	X	E	T	A	R	T	I	N	C	N	I	Z	D	E	G	Z	R
B	D	E	F	H	J	Z	I	N	C	N	I	T	R	I	T	E	L	O	N	A
P	R	T	V	Z	B	D	F	E	T	A	R	T	I	N	M	U	I	R	A	B
H	J	L	N	P	C	A	L	C	I	U	M	F	L	U	O	R	I	D	E	A
E	T	A	N	O	B	R	A	C	M	U	I	S	S	A	T	O	P	Y	Z	F
R	T	V	X	E	D	I	M	O	R	B	N	E	G	O	R	D	Y	H	N	U



*Write the formulas for the following covalent molecules*

- 1) antimony tribromide \_\_\_\_\_
- 2) hexaboron monosilicide\_\_\_\_\_
- 3) chlorine dioxide \_\_\_\_\_
- 4) sulfur monoiodide \_\_\_\_\_
- 5) iodine pentafluoride \_\_\_\_\_
- 6) dinitrogen trioxide \_\_\_\_\_
- 7) ammonia \_\_\_\_\_
- 8) phosphorus triiodide \_\_\_\_\_

*Write the names for the following covalent molecules*

- 9)  $\text{P}_4\text{S}_5$  \_\_\_\_\_
- 10)  $\text{O}_2$  \_\_\_\_\_
- 11)  $\text{SeF}_6$  \_\_\_\_\_
- 12)  $\text{Si}_2\text{Br}_6$  \_\_\_\_\_
- 13)  $\text{SCl}_4$  \_\_\_\_\_
- 14)  $\text{CH}_4$  \_\_\_\_\_
- 15)  $\text{B}_2\text{Si}$  \_\_\_\_\_
- 16)  $\text{NF}_3$  \_\_\_\_\_

## Mixed Naming Practice – both ionic and covalent

DON'T FORGET TO CHECK IF IT IS IONIC OR COVALENT FIRST!!!! It changes how you name them! Also, don't forget to include Roman Numerals if it is a transition metal!

*Name the following chemical compounds:*

- 1) NaBr \_\_\_\_\_
- 2) Ca(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> \_\_\_\_\_
- 3) P<sub>2</sub>O<sub>5</sub> \_\_\_\_\_
- 4) Ti(SO<sub>4</sub>)<sub>2</sub> \_\_\_\_\_
- 5) FePO<sub>4</sub> \_\_\_\_\_
- 6) K<sub>3</sub>N \_\_\_\_\_
- 7) SO<sub>2</sub> \_\_\_\_\_
- 8) CuOH \_\_\_\_\_
- 9) Zn(NO<sub>2</sub>)<sub>2</sub> \_\_\_\_\_
- 10) V<sub>2</sub>S<sub>3</sub> \_\_\_\_\_

*Write the formulas for the following chemical compounds:*

- \*WAIT!\*** 11) silicon dioxide \_\_\_\_\_
- Don't do  
these yet,** 12) nickel (III) sulfide \_\_\_\_\_
- I will tell  
you when  
to come  
back and  
do this  
last part!** 13) manganese (II) phosphate \_\_\_\_\_
- 14) silver acetate \_\_\_\_\_
- 15) diboron tetrabromide \_\_\_\_\_
- 16) potassium carbonate \_\_\_\_\_
- 17) ammonium oxide \_\_\_\_\_
- 18) carbon tetrachloride \_\_\_\_\_